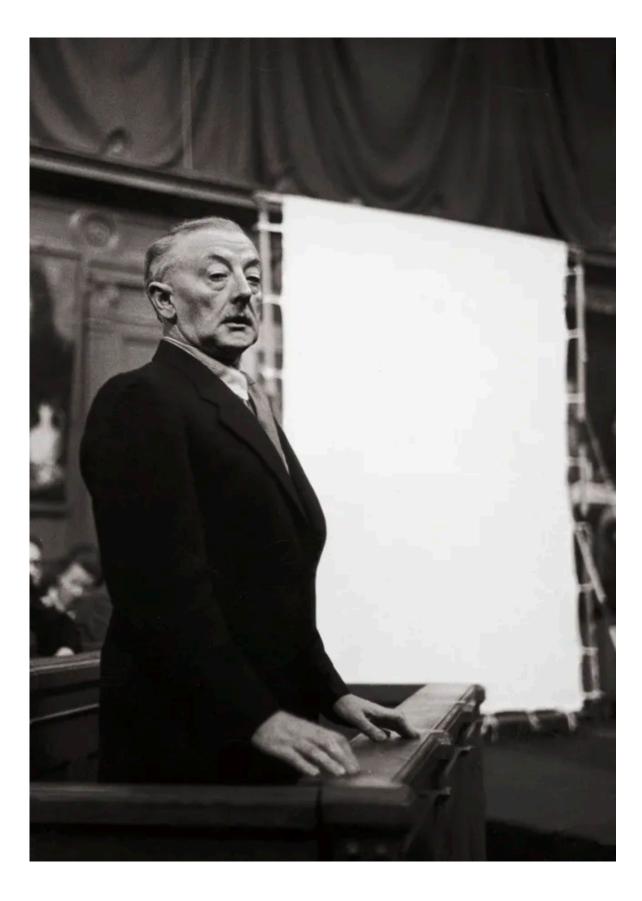
# Van Meegeren art forgeries



From M. Braun, "Differential Equations and their applications", 4th ed., Springer

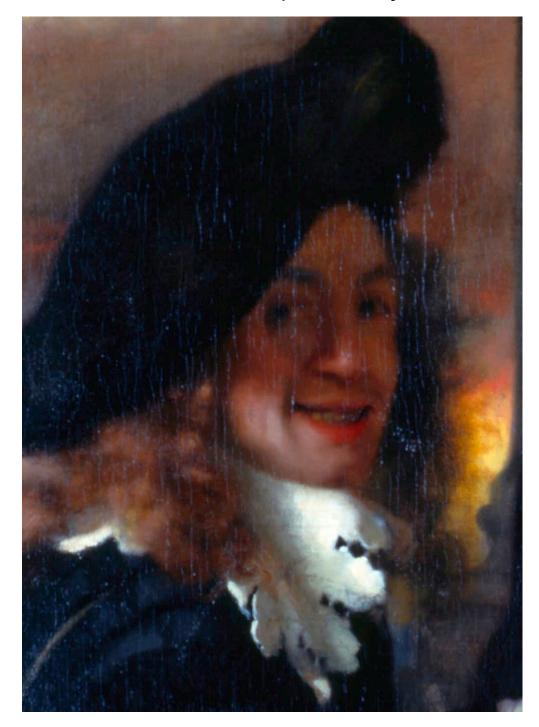




#### Disciples at Emmaus, by Johannes Vermeer (1632-1675)

• "the masterpiece of Johannes Vermeer of Delft!", according to Abraham Bredius • sold to the Rembrandt society for 170,000\$ (about \$4M today) in 1937

> detail of the painting The Procuress, believed to be a self portrait by Vermeer

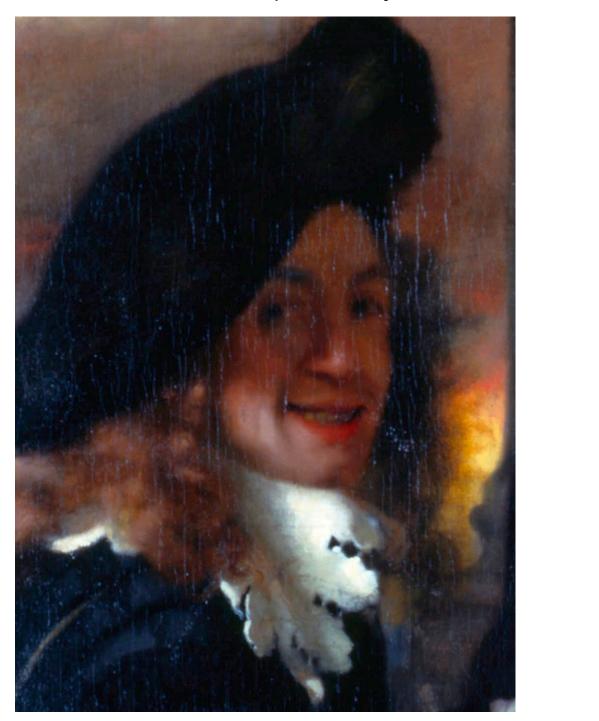






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Girl with a pearl earring



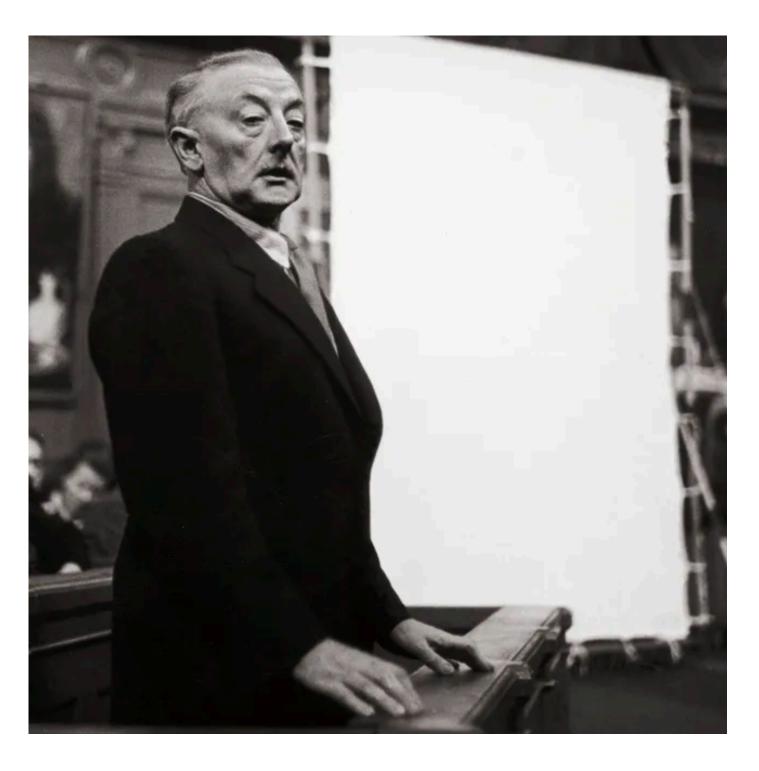


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Dutch painter Got arrested at the end of World War II for collaboration with the Nazis

To defend himself, he claimed to have fooled the Germans by selling them lots of fake paintings of famous Dutch painters:

in particular, Woman taken in adultery and Disciples of Emmaus



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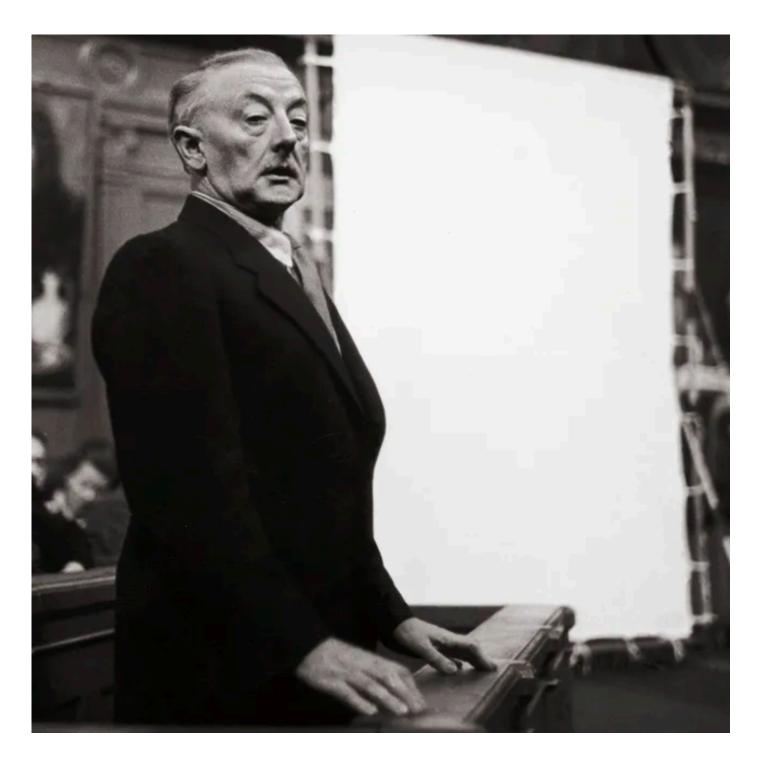
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Woman taken in adultery and other less famous paintings were easily recognized as a fake.Van Meegeren was convicted to one year of prison of forgery and died shortly after.However, the question about authenticity on *Disciples of Emmaus* remained...

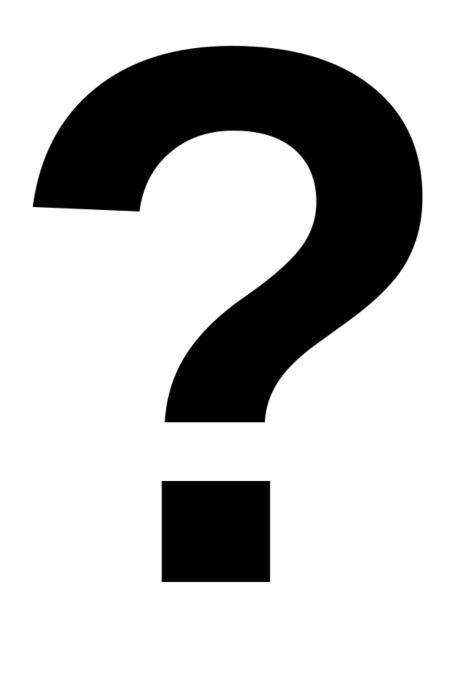




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The answer was found by scientists at Carnegie Mellon University in 1967



Radioactive decay:

Atoms of certain "radioactive" elements are unstable and within a given time period a proportion of the atoms spontaneously disintegrates to form atoms of a new element.

Radioactivity of a substance is directly proportional to the number of atoms of the substance present.

$$\frac{dy}{dt} = -\lambda y$$
$$y(t_0) = y_0$$

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Separable equation: solution is

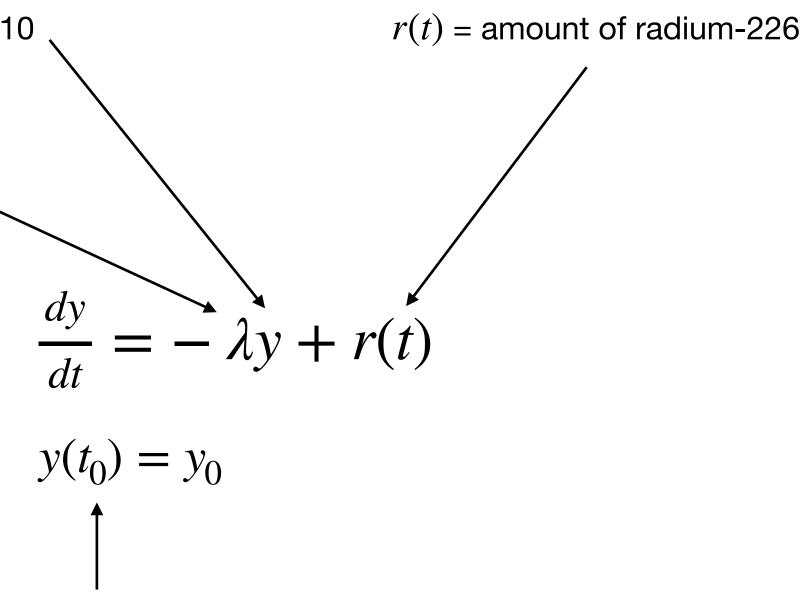
 $\frac{dy}{dt} = -\lambda y$  $y(t_0) = y_0$ 

 $y(t) = y_0 e^{-\lambda(t-t_0)}$ 

all paintings contain a pigment (white lead) that has a small amount of lead-210 and of radium-226 (two radioactive elements) disintegration of the lead-210 is exactly balanced by the disintegration of the radium

y(t) = amount of lead-210

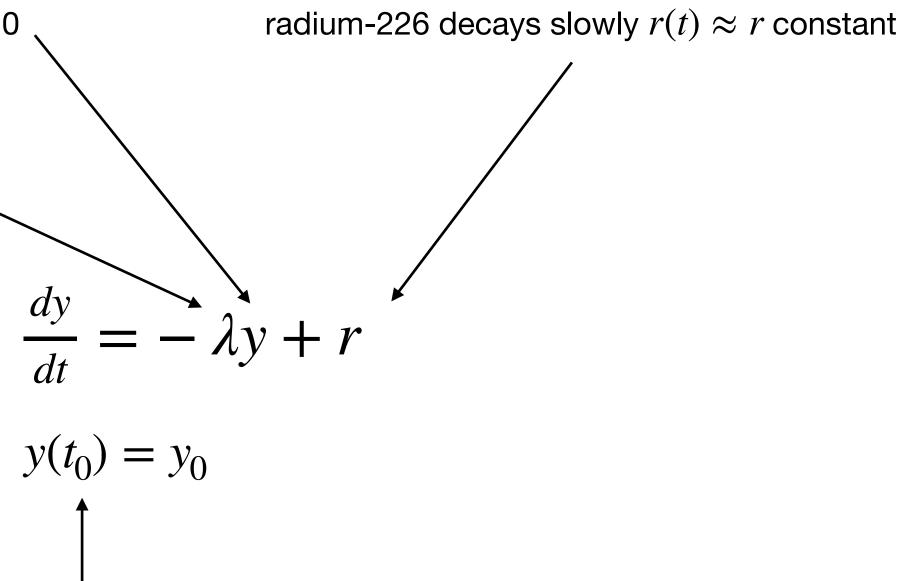
$$\lambda = \frac{\ln 2}{22}$$
 = rate of decay of lead-210



 $t_0$  = time of creation of the painting

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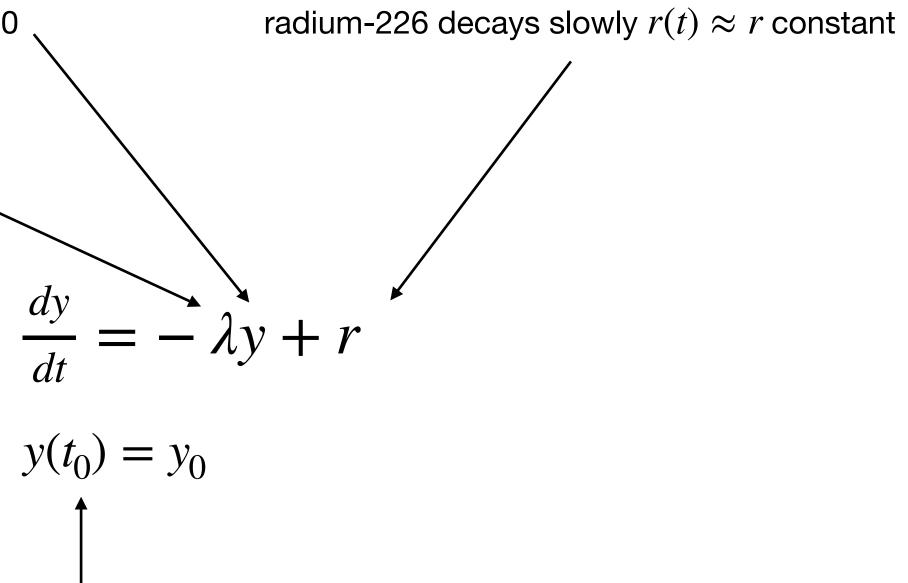
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Linear equation: solution is

$$y(t) = \frac{r}{\lambda}(1 - e^{-\lambda(t - t_0)}) + y_0 e^{-\lambda(t - t_0)}$$



 $t_0$  = time of creation of the painting

 $y(t) = \frac{r}{\lambda}(1 - e)$ 

$$e^{-\lambda(t-t_0)}) + y_0 e^{-\lambda(t-t_0)}$$

 $y(t) = \frac{r}{\lambda}(1 - t)$ 

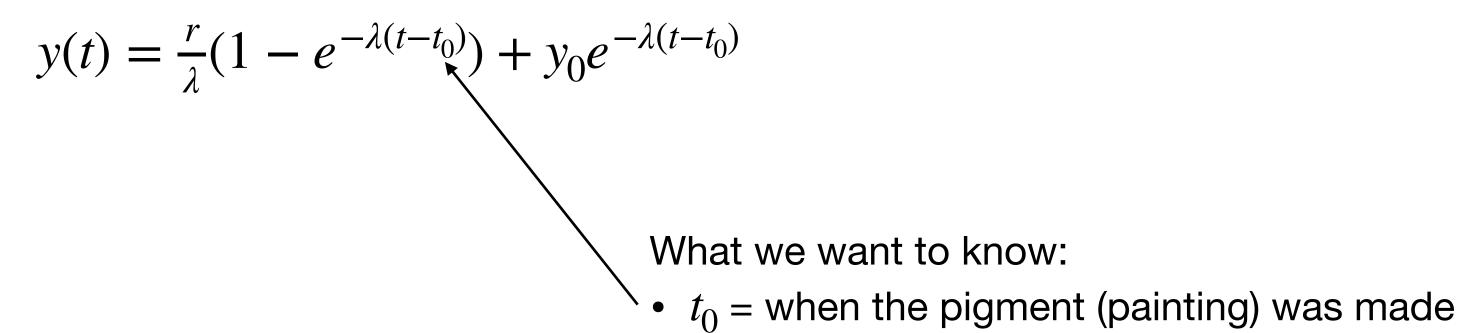
What we know:

- $\lambda$  = rate of decay of lead-210
- r =amount of radium-266
- y(t) = amount of lead-210 in the present day

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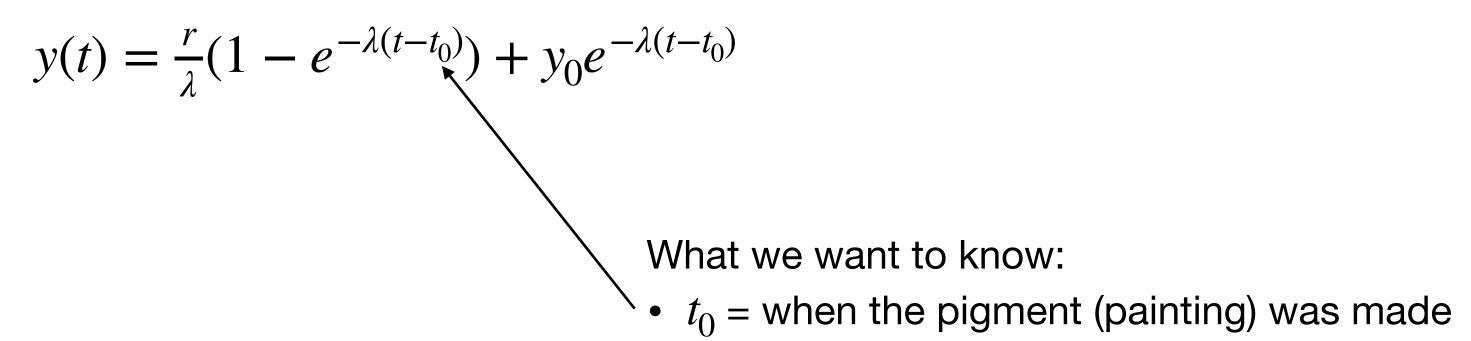


What we know:

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What we're missing:

(but we can estimate that!)



•  $y_0$  = amount of lead-210 at the time of making

Assume the painting is indeed old:  $t - t_0 \approx 300$  years

Rewrite the solution as

$$\begin{split} \lambda y_0 &= \lambda y(t) e^{\lambda(t-t_0)} - r(e^{\lambda(t-t_0)} - 1) \\ &= \lambda y(t) e^{300\lambda} - r(e^{300\lambda} - 1) \\ &= 8.5 e^{300\frac{\ln 2}{22}} - 0.8(e^{300\frac{\ln 2}{2}} - 1) \\ &= 98,050 \end{split}$$

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 $\lambda y_0 = \lambda y(t) e^{\lambda(t-t)}$  $= \lambda y(t) e^{300\lambda}$  $= 8.5 e^{300\frac{\ln 2}{22}}$ = 98,050

This quantity shouldn't be bigger than 30,000 (by comparing with old rocks)

$$r^{-t_0} - r(e^{\lambda(t-t_0)} - 1)$$
  
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## This painting is fake!!!

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